

STEREOSELECTIVE SYNTHESIS OF THE BASIC SKELETON OF  
APHIDICOLAN DITERPENES SYNTHETIC APPROACH TO APHIDICOLIN

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Summary The basic skeleton of aphidicolan-type diterpenes was stereoselectively synthesized via thermolysis of a benzocyclobutene

Aphidicolin (1), isolated from a culture of *Cephalosporium aphidicola* Petch by Hesp<sup>1</sup>, has a unique carbon-framework and has been shown to display antiviral activity. Two elegant total syntheses of 1 have recently been published by Trost<sup>2</sup> and McMurry<sup>3</sup>. Ireland<sup>4</sup> has also reported an interesting synthetic approach to 1. We here wish to report a stereoselective synthesis of the basic aphidicolan skeleton by an alternative approach which involves thermolysis of a benzocyclobutene as the key reaction.

The benzocyclobutene (2)<sup>5</sup> was alkylated with the ethylene acetal (3) to give the acetal (4) in 73 % yield. After deprotection of the carbonyl with acid, the aldehyde (5) was treated with 3-butenyl-1-magnesium bromide to afford the alcohol (6) which was converted to the corresponding ketone (7) in 71 % yield using pyridinium chlorochromate<sup>6</sup>. Spectral data [ $\nu_{\max}$  (CHCl<sub>3</sub>) 2235 (C≡N) and 1715 (C=O) cm<sup>-1</sup>,  $\delta$  (CDCl<sub>3</sub>) 2.22 (3H, s, Ar-CH<sub>3</sub>), 3.17 (1H, d, J=14 Hz, Ar- $\overset{\text{H}}{\underset{\text{H}}{\text{C}}}$ ), 3.67 (1H, d, J=14Hz, Ar- $\overset{\text{H}}{\underset{\text{H}}{\text{C}}}$ ), 3.83 (3H, s, OCH<sub>3</sub>), 5.00 - 5.30 (2H, m, -CH=CH<sub>2</sub>), 5.68 - 6.28 (1H, m, -CH=CH<sub>2</sub>), 6.85 (1H, br s, arom) and 6.91 (1H, br s, arom), m/e 283 (M<sup>+</sup>)] is consistent with structure 7. Thermolysis of the benzocyclobutene (7) was carried out in refluxing *o*-dichlorobenzene for 6 h to furnish the cyclized product (8) in 55 % yield, [ $\nu_{\max}$  (CHCl<sub>3</sub>) 2225 (C≡N) and 1705 (C=O) cm<sup>-1</sup>,  $\delta$  (CDCl<sub>3</sub>) 2.40 (3H, s, Ar-CH<sub>3</sub>), 3.82 (3H, s, OCH<sub>3</sub>), 6.80 (1H, br s, arom) and 6.95 (1H, d, J=2Hz, arom), m/e 283 (M<sup>+</sup>). Base treatment of the compound (8), followed by acid hydrolysis, gave the tetracyclic compound (9), m/e 284 (M<sup>+</sup>). The i.r. spectrum of 9 showed two carbonyl absorptions (at 1735 and 1710 cm<sup>-1</sup>) and no nitrile absorption. The stereochemical assignment was made on the basis of the n.m.r. spectrum [ $\delta$  (CDCl<sub>3</sub>) 2.20 (3H, s, Ar-CH<sub>3</sub>), 3.87 (3H, s, OCH<sub>3</sub>), 6.70 (1H, d, J=2Hz, arom) and 7.63 (1H, d, J=2Hz, arom)] which showed one low field aromatic proton. From an examination of molecular models it is seen that the aphidicolan-type stereoisomer (9) can be depicted as in (A) or (B) of Fig 1. In formula A, one aromatic proton is clearly deshielded by a carbonyl group. As this effect was observed in the n.m.r. spectrum of the above product the stereochemistry is assigned as in formula 9. The same effect has been observed with the A-ring aromatic iso-drimenon ring system<sup>7</sup>. The conversion of 9 to aphidicolin (1) is now under progress in this laboratory.

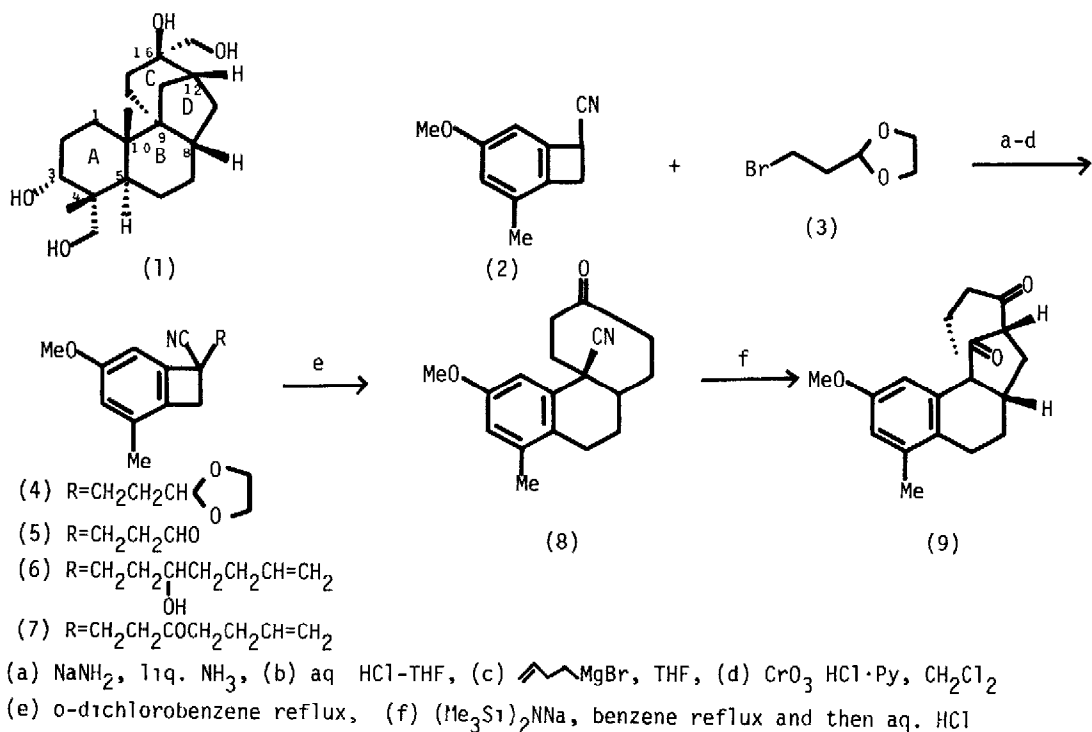
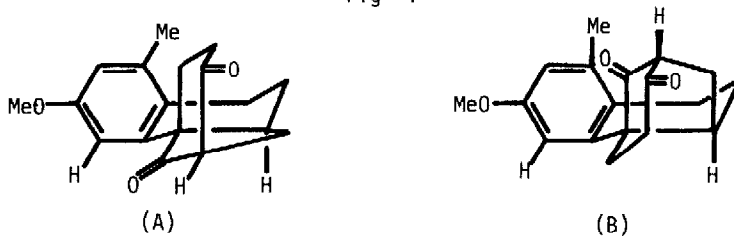


Fig 1



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